Investigation of Activation Energy and Reaction Mechanism in the Synthesis of Li?SiO?

Abstract

Lithium orthosilicate (Li?SiO?) has gained interest as a tritium breeder material in fusion reactors. It is a viable candidate for this purpose, as it can produce tritium when exposed to neutron irradiation. Li?SiO? has strong thermal stability, excellent chemical compatibility with other reactor materials, and good tritium release characteristics. Furthermore, Li?SiO? has a high lithium density, which improves tritium breeding efficiency. These qualities make it an essential component in the blanket of fusion reactors, where it plays a critical role in ensuring a continuous tritium supply for reactor operation.

Li?SiO? is typically synthesized through a solid-state reaction with lithium carbonate (Li?CO?) and silicon dioxide (SiO?) as a raw materials. Several factors influence this reaction, including temperature, reaction duration, reactant particle size, and the presence of contaminants. Understanding the reaction kinetics of Li?SiO? synthesis is critical for refining the synthesis process. Although Li?SiO? is crucial in fusion technology, little research has been conducted on its reaction kinetics. A detailed investigation of reaction processes, rate constants, and activation energies is required to design efficient and scalable manufacturing procedures. This study intends to investigate the reaction kinetics of Li?SiO? synthesis by thermal analysis. Student will get hands on experience with TG-DSC for thermal analysis, ball-milling for grinding, and high-temperature furnaces for Li4SiO4 synthesis.

Discipline Required: Chemical Engg. (Any sub-major is also acceptable)

Academic Project Requirements:

- 1) Required No. of student(s) for academic project: 1
- 2) Name of course with branch/discipline: M.E./M.Tech Other
- 3) Academic Project duration:
- (a) Total academic project duration: 38 Weeks
- (b) Student's presence at IPR for academic project work: 3 Full working Days per week

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